

## SYNTHESIS OF SOME NEW HETEROCYCLIC SCHIFF BASES, 4-THIAZOLIDINONES AND 2-AZETIDINONES AS AN ANTIBACTERIAL AND ANTIFUNGAL AGENT

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### ABSTRACT

Some new heterocyclic Schiff bases (**1a-h**) were synthesized from 2-amino-6-substituted benzothiazoles. Further these heterocyclic Schiff bases were converted into 4-thiazolidinones (**2a-h**) and 2-azetidinones (**3a-h**) by the action of mercaptoacetic acid and chloroacetyl chloride respectively. The biological screening data of the synthesized compounds were also presented.

**Key words :** Schiff bases, 4-Thiazolidinone, 2-Azetidinone, Antibacterial, Antifungal.

### INTRODUCTION

Heterocyclic compounds of Schiff bases like 4-thiazolidinones and 2-azetidinones are reported as anticancer<sup>1-3</sup>. Schiff bases possess diversified biological applications<sup>4,5</sup>. Various 4-thiazolidinones show a variety of pharmacological activities<sup>6,7</sup>. Moreover, compounds containing 2-azetidinone ring system are shown to possess marked biological activities<sup>8-11</sup>. Various 4-thiazolidinones inhibit the bacterial enzyme in biosynthesis of polymers<sup>12</sup>. All these observations and the essential role of heterocyclic compounds prompted us to synthesize Schiff bases (**1a-h**), 4-thiazolidinones (**2a-h**) and 2-azetidinones (**3a-h**).

2-hydroxy-3-iodo-5-bromo/chloro benzaldehyde on condensation with 2-amino-6-substituted benzothiazole furnished the Schiff bases (**1a-h**). These Schiff bases on cyclo condensation with mercapto acetic acid in dioxane and in presence of anhydrous zinc chloride afforded 4-thiazolidinones (**2a-h**). Schiff bases (**1a-h**) on reaction with chloroacetyl chloride in dioxane and in presence of triethylamine yield 2-azetidinones (**3a-h**) (Scheme 1). Further the structure of compounds were deduced on the basis of elemental analysis and spectral data (IR and <sup>1</sup>H NMR).

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### Antimicrobial Activity

The compounds synthesized were screened for their antibacterial activity using *Escherichia coli* (EC), *Salmonella typhi* (ST) and *Salmonella dysenteriae* (SD) as bacteria. The activities of these compounds were tested using disc diffusion method<sup>13</sup> at 150 ppm. concentration using 5 mm filter paper disc. Tetracycline an antibiotic was used as a standard for comparison. The area of inhibition of zone was measured. Compounds **1a**, **1d**, **1h**, **2a**, **2g**, **3a** and **3d** showed good antibacterial activity. Remaining other compounds showed moderate to less activity.

The antifungal activity was tested against the fungal species *Aspergillus niger* (AN), *Penicillium notatum* (PN) and *Alternaria tenuissiana* (AT) at 150 ppm concentration. The antifungal data revealed the compounds **1d**, **1e**, **2d**, **2h**, **3b** and **3h** to be moderately active against the fungi. Other compounds were found to be less active against the same fungal species.

### EXPERIMENTAL

Melting points were determined in open capillaries in a liquid paraffin bath and are uncorrected. Purity of compounds was checked by TLC. IR spectra were recorded in nujol on Perkin-Elmer-237 spectrophotometer. <sup>1</sup>H NMR were recorded in CDCl<sub>3</sub> on a Perkin-Elmer R-32 spectrometer using TMS as internal standard (Chemical shift are given in δ ppm).

#### Preparation of 2-N-(2-hydroxy-3-iodo-5-bromo benzylidene)-6-methoxy benzothiazole (**1e**)

A mixture of 2-hydroxy-3-iodo-5-bromo benzaldehyde (0.001 mole) and 2-amino-6-methoxy benzothiazole (0.001 mole) were dissolved in ethyl alcohol (25 mL). One drop of acetic acid was added and was refluxed for 2 hrs. The resultant solution was cooled and poured in cold water. The separated solid was filtered, crystallized from ethyl alcohol to give **1e**.  $\nu_{\max}$  1630 (C=N) and 1590 cm<sup>-1</sup> (C=C). <sup>1</sup>H NMR: δ 4.2 (s, 3H, OCH<sub>3</sub>), 6.9–7.4 (m, 5H, Ar-H), 6.7 (s, 1H, CH=) and 7.5 (s, 1H, Ar-OH). Similarly other compounds were also prepared (Table 1).

#### Preparation of 2-(2-hydroxy-3-iodo-5-bromo phenyl)-3-(6-methoxybenzothiazoly)-4-thiazolidinone (**2e**)

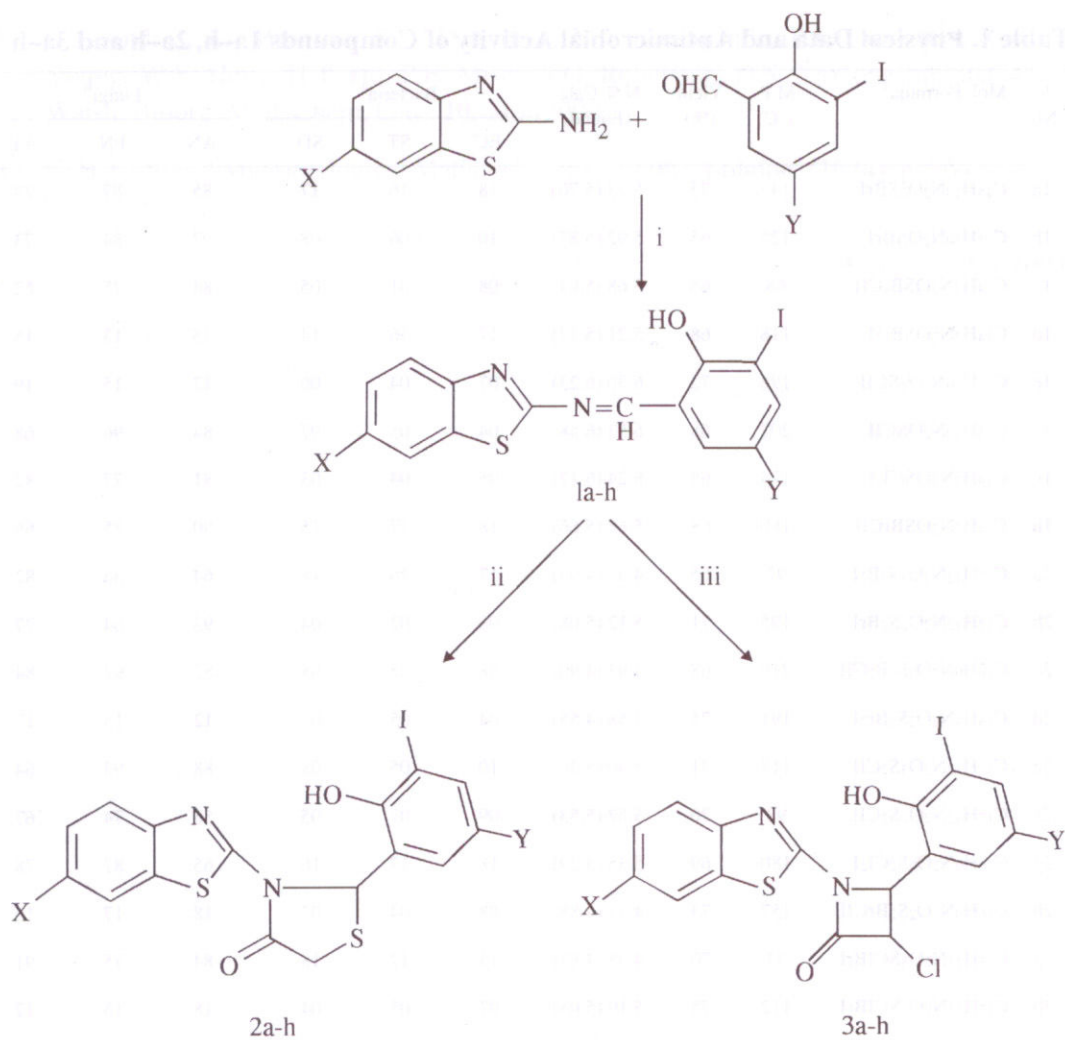
A mixture of compound **1e** (0.001 mole) and mercapto acetic acid (0.001 mole) were dissolved in dioxane (20 mL). Pinch of anhydrous zinc chloride was added and then refluxed for 8 hrs. Separated solid was filtered, washed with sodium bicarbonate solution and then crystallized from ethyl alcohol to give **2e**.  $\nu_{\max}$  1665 (C=O) and 1580–1630 cm<sup>-1</sup> (C=C).

<sup>1</sup>H NMR: δ 3.9 (s, 3H, OCH<sub>3</sub>), 4.5 (s, 2H, CH<sub>2</sub>S), 6.5 (s, 1H, N-CH), 7.0–7.9 (m, 5H, Ar-H) and 8.9 (s, 1H, Ar-OH). Similarly other compounds were also prepared (Table 1).

**Table 1. Physical Data and Antimicrobial Activity of Compounds 1a–h, 2a–h and 3a–h**

S. No.	Mol. Formula	M.P. (°C)	Yield (%)	N % Calc. (Found)	Bacteria <sup>a</sup>			Fungi <sup>b</sup>		
					EC	ST	SD	AN	PN	AT
1a	C <sub>15</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub> SBrI	147	73	5.73 (5.70)	18	16	17	85	87	78
1b	C <sub>15</sub> H <sub>10</sub> N <sub>2</sub> OSBrI	125	65	5.92 (5.87)	10	06	08	97	84	73
1c	C <sub>14</sub> H <sub>7</sub> N <sub>2</sub> OSBrCII	68	65	5.68 (5.64)	08	01	05	84	75	82
1d	C <sub>14</sub> H <sub>7</sub> N <sub>2</sub> OSBr <sub>2</sub> I	118	68	5.21 (5.17)	17	16	14	15	13	15
1e	C <sub>15</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub> SCII	198	72	6.30 (6.23)	07	04	06	17	15	19
1f	C <sub>15</sub> H <sub>10</sub> N <sub>2</sub> OSCI	202	68	6.53 (6.48)	04	05	07	84	96	68
1g	C <sub>14</sub> H <sub>7</sub> N <sub>2</sub> OSCl <sub>2</sub> I	174	65	6.24 (6.17)	05	04	03	81	77	82
1h	C <sub>14</sub> H <sub>7</sub> N <sub>2</sub> OSBrCII	180	68	5.68 (5.66)	18	17	15	90	85	66
2a	C <sub>17</sub> H <sub>12</sub> N <sub>2</sub> O <sub>3</sub> S <sub>2</sub> BrI	92	65	4.97 (4.93)	17	16	18	64	56	82
2b	C <sub>17</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> S <sub>2</sub> BrI	195	71	5.12 (5.08)	06	02	04	93	64	77
2c	C <sub>16</sub> H <sub>9</sub> N <sub>2</sub> O <sub>2</sub> S <sub>2</sub> BrCII	210	68	4.93 (4.90)	08	05	08	82	97	84
2d	C <sub>16</sub> H <sub>9</sub> N <sub>2</sub> O <sub>2</sub> S <sub>2</sub> Br <sub>2</sub> I	191	75	4.58 (4.55)	04	05	07	12	15	17
2e	C <sub>17</sub> H <sub>12</sub> N <sub>2</sub> O <sub>3</sub> S <sub>2</sub> CII	118	71	5.40 (5.36)	10	05	08	88	93	64
2f	C <sub>17</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> S <sub>2</sub> CII	93	70	5.57 (5.53)	09	07	05	56	74	67
2g	C <sub>16</sub> H <sub>9</sub> N <sub>2</sub> O <sub>2</sub> S <sub>2</sub> Cl <sub>2</sub> I	180	69	5.35 (5.29)	18	17	16	65	87	78
2h	C <sub>16</sub> H <sub>9</sub> N <sub>2</sub> O <sub>2</sub> S <sub>2</sub> BrCII	157	73	4.93 (4.88)	08	04	07	18	17	16
3a	C <sub>17</sub> H <sub>11</sub> N <sub>2</sub> O <sub>3</sub> SCIBrI	93	70	4.95 (4.87)	15	17	18	84	75	91
3b	C <sub>17</sub> H <sub>11</sub> N <sub>2</sub> O <sub>2</sub> SCIBrI	112	75	5.10 (5.03)	07	05	04	18	15	17
3c	C <sub>16</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub> SCI <sub>2</sub> BrI	108	65	4.91 (4.86)	04	03	01	48	87	64
3d	C <sub>16</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub> SCIBr <sub>2</sub> I	119	73	4.56 (4.49)	19	17	18	85	47	63
3e	C <sub>17</sub> H <sub>11</sub> N <sub>2</sub> O <sub>3</sub> SCI <sub>2</sub> I	110	73	5.38 (5.27)	05	04	03	84	76	85
3f	C <sub>17</sub> H <sub>11</sub> N <sub>2</sub> O <sub>2</sub> SCI <sub>2</sub> I	75	68	5.55 (5.48)	07	05	04	76	64	53
3g	C <sub>16</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub> SCI <sub>2</sub> I	120	67	5.33 (5.29)	08	04	06	58	47	33
3h	C <sub>16</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub> SCI <sub>2</sub> BrI	109	65	4.91 (4.87)	10	08	04	18	17	15
				Tetracycline	20	21	20	–	–	–
				Greseofulvin	–	–	–	15	20	15

<sup>a</sup> Zone of inhibition in mm, <sup>b</sup> % of germination after 12 hr.

1,2 & 3a,e X = OCH<sub>3</sub>

1,2 &amp; 3a-d, Y = Br

b,f X = CH<sub>3</sub>

1,2 &amp; 3e-h, Y = Cl

c,g X = Cl

d,h X = Br

Reagents (i) EtOH (ii) HS - CH<sub>2</sub> - COOH, ZnCl<sub>2</sub>/Dioxane (iii) Cl - CH<sub>2</sub> - COCl, Et<sub>3</sub>N/Dioxane**Scheme - 1**

### Preparation of 1-(6-methoxybenzothiazolyl)-3-chloro-4-(2-hydroxy-3-iodo-5-bromo phenyl)-2-azetidinone (3e)

A mixture of compound **1e** (0.001 mole) and triethylamine (0.003 mole) were dissolved in dioxane (25 mL). Chloroacetyl chloride (0.0012 mole) was added dropwise at 10°C. The reaction mixture was stirred for 6 hrs. Half of the solvent was removed by distillation and then cooled. Separated out solid was crystallized from chloroform to give **3e**.  $\nu_{\max}$  1780 (C=O) and 1666, 1599  $\text{cm}^{-1}$  (C=C).  $^1\text{H NMR}$ :  $\delta$  3.8 (s, 3H,  $\text{OCH}_3$ ), 4.2 (d, 1H, N-CH), 4.6 (d, 1H, CH-Cl), 6.7-7.7 (m, 5H, Ar-H) and 9.2 (s, 1H, Ar-OH). Similarly other compounds were also prepared (Table 1).

All compounds gave satisfactory C, H and N analysis.

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