



Trade Science Inc.

Environmental Science

An Indian Journal

Current Research Paper

ESAIJ, 6(6), 2011 [347-350]

S/S Technology acts as a tool for protecting the environment from hazardous pollutants

Rana Pratap Singh, Rubina Chaudhary*, Shukti Tomar

School Of Energy And Environmental Studies, Devi Ahilya University, Indore, (INDIA)

E-mail : rubina_chaudhary@yahoo.com

Received: 21st August, 2011 ; Accepted: 21st September, 2011

ABSTRACT

In stabilization/ solidification (S/S) of solid wastes by means of adding cementitious binders, like lime and cement the toxic constituents present in the waste form are physically as well as chemically “fixed”, that is, their mobility is significantly reduced so as to minimize the threat to the environment and also ensure compliance with existing regulatory standards. This paper not only studies the role of stabilization of waste from some industries by adding binders like lime, cement and flyash in various ratios but also analysed its leachability and cost effective ratio.

© 2011 Trade Science Inc. - INDIA

KEYWORDS

Solidification/ Stabilization;
Binder;
Hazardous Waste;
Leachability.

INTRODUCTION

Protecting the environment from hazardous pollutants associated with waste generation and disposal is a major concern in today's heavily industrialized world. Thus far, various technologies have been developed which transform hazardous wastes to nontoxic, or reduce the potential release of toxic species into the environment. One such promising technology is stabilization/ solidification (S/S) of solid wastes by means of adding cementitious binders, like lime and cement. During S/S applications, the toxic constituents present in the waste form are physically as well as chemically “fixed”, that is, their mobility is significantly reduced so as to minimize the threat to the environment and also ensure compliance with existing regulatory standards. This paper not only studies the role of stabilization of waste from some industries by adding binders like lime,

cement and flyash in various ratio but also check its leachability and most cost effective ratio.

MATERIAL AND METHOD

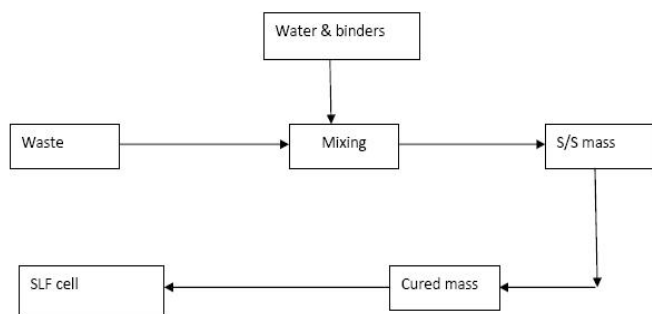
Physico-chemical characterization of waste

Waste taken for study was from three industries i.e. Metal Industry, Automobile Industry and Steel industry. The results indicate us that whether the waste is to be treated before landfilling or is incineration. The heavy metal concentration was also analysed for these waste and accordingly subjected to stabilization (TABLE 2&3).

Mechanism of stabilization:

To develop stabilization recipe for the specific waste series of experiment with the different percentage of stabilization were performed. A known quantity of waste

Current Research Paper



Flow chart of mechanism of stabilization

to be stabilized is taken in 200ml beaker. If a waste was solid or semi solid sufficient amount of water was added to make free flowing waste slurry then known quantity of absorbent material was added like fly ash until the thick slurry stop appearing. Then known quantity of lime, sodium silicate and cement was added as waste characteristics and mixed well. If liquid rises within 30 minutes than again add absorbent material and leave the beaker for 24 hours to cure and then perform the TCLP test for analysing its leachability. (TABLE 2.3)

RESULT AND DISCUSSION

Physico-chemical characterization

Waste from all the industry were subjected to various physical and chemical test for various parameter like physical test, calorific value, and heavy metal concentration etc. (TABLE 2 & 3)

From the TABLE 2, it is seen that all industrial waste

TABLE 1 : Preparation of Samples for Stabilization of waste

Samples	Sample Weight (gm)	Fly Ash (%)	Cement (%)	Lime (%)	Sodium Sulphite (%)
1	30	16	8	5	0.5
2	30	10	10	5	0.5
3	30	20	10	5	0.5

has calorific value below 2500 cal/gram, so these waste were not suitable for incineration and they required treatment before disposal.

TABLE 3 : Heavy Metals Concentration

Parameter	Method	Result			TCLP Test Limits as per CPCB Guidelines HAZWAMS /32/ 2005-2006
		TCLP (mg/l) Metal Industry	TCLP (mg/l) Automobile Industry	TCLP (mg/l) Steel Industry	
Cadmium as Cd	WLT Method	BDL	0.01	BDL	1.0
Chromium as Cr	WLT Method	BDL	BDL	BDL	5.0
Copper as Cu	WLT Method	BDL	BDL	0.1	3.0
Iron as Fe	WLT Method	BDL	BDL	BDL	30.0
Lead as Pb	WLT Method	7.0	0.1	BDL	5.0
Manganese as Mn	WLT Method	BDL	15.0	0.7	Not Specified
Nickel as Ni	WLT Method	0.3	BDL	7.9	3.0
Zinc as Zn	WLT Method	0.1	13.0	0.2	15.0

TABLE 2 : Physico-chemical characterization

Parameter	Unit	Method	Result			CPCB Guidelines HAZWAMS/32/2005-2006
			Metal Industry	Automobile Industry	Steel Industry	
Physical State	-	SW 846	Solid	Solid	Semi Solid	Not Specified
Colour	-	SW 846	Black	Grey	Grey	Not Specified
Texture	-	SW 846	Solid	Dry Powder	Wet Lumps	Not Specified
Specific Gravity	g/cm ³	ASTM-D 5057-90	1.4	0.7	1.9	Not Specified
Calorific Value	cal/g	IS:1350 Part II-1970	275.0	545.0	2311.0	<2500 cal/g
Flash Point	°C	SW 846 1020 A	>60 °C	>60 °C	>60 °C	> 60 °C
LOD @ 105 °C	%	APHA 2540	0.6	6.97	45.3	Not Specified
LOI @ 550 °C	%	APHA 2540	2.7	10.7	17.3	< 20 %
pH (at room temperature)	-	SW 8469045 C	6.7	7.0	3.8	>4 to <12

From the TABLE 3, it is seen that in metal industry only lead concentraion is above the TCLP permissible limit and rest heavy metal within TCLP permissible limit, likewise in automobile industry manganese and zinc concentration is above TCLP limit, also in Steel industry Nickel concentration is above TCLP limit and rest are either below detectible limit or within the TCLP limit.

Satbilization of waste

Waste from the three industries gets sucessfully stablized as shown by the results obtained by TCLP test. On the basis of cost of each binders and quatity used in for stabilizing waste and also on the basis of TCLP test result cost effective sample were sorted out.

TABLE 4 : Metal Industry

Metal Industry	
Before Stabilization	After Stabilization
Pb concentration 7.0 mg/l	Pb concentration
	Sample 1 0.5 mg/l
	Sample 2 0.2 mg/l
	Sample 3* BDL

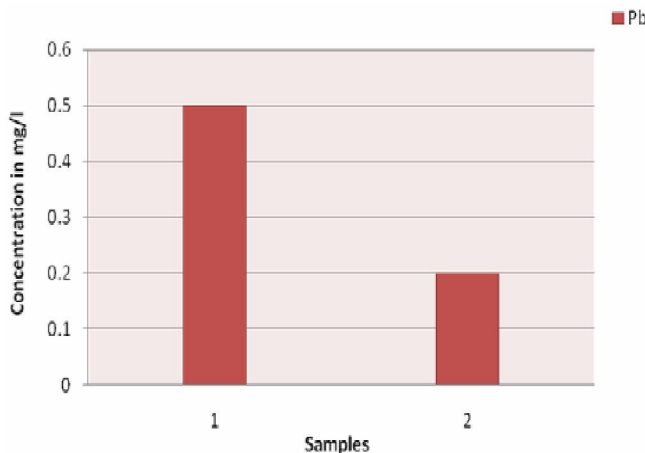


Figure 1 : Pb Concentration After Treatment

*Sample 3 is BDL in Figure

From the Figure 1 and TABLE 4, it was found that Sample no. 1, 2 &3 are successfully stabilized and the concentrations in all three samples are within the permissible limit. Sample no.1 is more cost effective than other samples.

From the Figure 2 and TABLE 5, it was found that Sample no. 1&3 are successfully stablized and the concentration of zinc and manganese are within permissible limit. But on cost effective point of view, the Sample no. 1 is best.

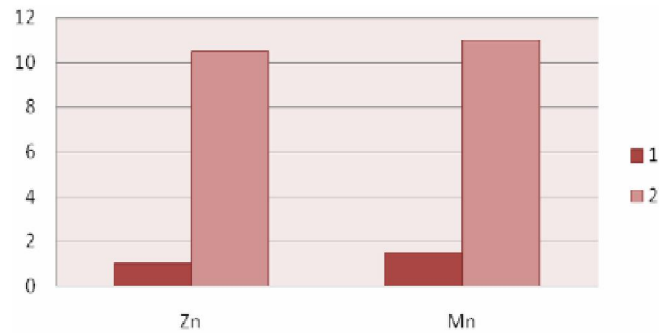


Figure 2 : Zn & Mn Concentration After Treatment

*Sample 3 is BDL in Figure

Table 5 : Auto mobile industry

Auto mobile Industry				
Before Stabilization		After Stabilization		
Zn concentration	Mn concentration		Zn concentration	Mn concentration
25mg/l	13mg/l	Sample 1	1.0 mg/l	1.5 mg/l
		Sample 2	10.5mg/l	11.0mg/l
		Sample 3*	BDL	BDL

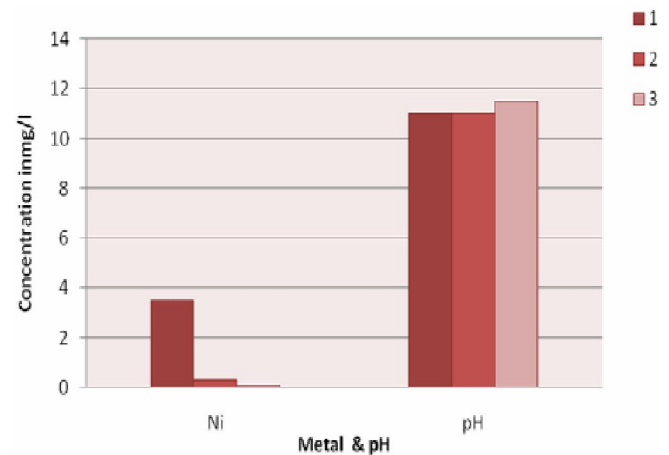


Figure 3 : Ni Concentration & pH After Treatment

TABLE 6 : Steel Industry

Steel Industry				
Before Stabilization		After Stabilization		
Ni concentration	pH		Ni concentration	pH
7.9 mg/l	3.8	Sample 1	3.5 mg/l	11.0
		Sample 2	0.3mg/l	11.0
		Sample 3	0.1mg/l	11.5

From the Figure 3 and TABLE 5, it was found that Sample no. 2&3 are successfully stabilized and not only nickel concentration falls within permissible limit but even pH also gets higher. But on cost effective point of view,

Current Research Paper

the Sample no. 2 is found best.

CONCLUSION

The physico-chemical parameters had shown that waste was not to be incinerated and need treatment before disposal. The waste of different Industry was completely stabilised when mixed with the different proportion of binders. It is also seen that flyash, lime and cement are very effective binders. Ratio of binders used sample one is more environment friendly and cost beneficial for metal industries and auto mobile industry but for steel industry sample two is found to be more environment friendly and cost beneficial.

The stabilised waste after landfill does not affected the quality of Soil, Ground Water & Air. It is also recommended that the stabilized waste can be used for making concrete, bricks, blocks etc. for green construction material.

REFERENCES

- [1] Albinas Gailius, Bozena Vacenovska, Rostislav Drochytka; *Materials Science (MEDŽIA GOTYRA)*, **16(2)**, (2010).
- [2] In-Ho Yoon, Deok Hyun Moon, Kyoung-Woong Kim, Keun-Young Lee, Ji-Hoon Lee, Min Gyu Kim; *Journal of Environmental Management*, **91(11)**, 2322-2328 (2010).
- [3] Dimitris Dermatas, Xiaoguang Meng; *Engineering Geology*, **70(3-4)**, 377-394 (2003).
- [4] Smita Badur, Rubina Chaudhary; *Review of Advance Material Sciences*, **17**, 42-61 (2008).
- [5] Divya Khale, Rubina Chaudhary; *Journal of Materials Science*, **42**, 729-746, (2007).
- [6] Rachana Malviya, Rubina Chaudhary; *Journal of Hazardous Materials*, **B137**, 267-276 (2006).
- [7] Central Pollution Control Board, HAZWAMS/11/1998-99, "Guideline for Setting-Up of Operating Facility" Ministry of Environment and Forest, Government of India.
- [8] Central Pollution Control Board, HAZWAMS/17/2000-01, "Criteria for Hazardous Wastes Landfills" Ministry Of Environment and Forest, Government of India.
- [9] Central Pollution Board, HAZWAMS/20/2002-03, "Guidelines for Transportation of Hazardous Waste" Ministry of Environment and Forest, Government of India, New Delhi.
- [10] Central Pollution Control Board, HAZWAMS/20/2002-03, 'Manual for Design, Construction and Quality Control of Liners and Covers for Hazardous Waste Landfill', Ministry of Environment and Forest, Government of India, New Delhi.
- [11] M.Data, B.P.Parida, B.K.Guha, T.R.Srekrishnhnam; 'Industrial Solid Waste Management and Land Fill Practice', Naroha Publishing House, New Delhi, (1999).
- [12] United Nation Environmental Program, 'Treatment and Disposal Method for Waste Chemicals', international Register for Potentially Toxic Chemicals, Geneva, (1985).
- [13] www.cpcb.nic.in
- [14] www.envfor.nic.in
- [15] www.fao.org